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Key indicators

Single-crystal X-ray study T = 298 KMean σ (C–C) = 0.004 Å R factor = 0.063 wR factor = 0.124 Data-to-parameter ratio = 13.7

For details of how these key indicators were automatically derived from the article, see http://journals.iucr.org/e.

© 2006 International Union of Crystallography All rights reserved amino-2,3-dihydro-1,3,4-thiadiazole

4-(4-Methoxybenzoyl)-2,2-dimethyl-5-phenyl-

In the title compound, $C_{18}H_{19}N_3O_2S$, the thiadiazole ring adopts an envelope conformation. The crystal structure is stabilized by intermolecular $N-H\cdots O$ hydrogen bonds, forming one-dimensional chains parallel to the *b* axis. Received 23 January 2006 Accepted 13 February 2006

Comment

Thiadiazole derivatives such as 1-(1,3,4-thiadiazol-2-yl)-3-methylthio-6,7-dihydrobenzo[c]thiophen-4(5H)-one are known to exhibit high antibacterial activity against*Staphylococcus aureus*(Tehranchian*et al.*, 2005).



In the title compound, (I), the thiadiazole ring adopts an envelope conformation, with atom C8 at the flap position; C8 deviates from the mean plane through the other four atoms by 0.204 (3) Å. The dihedral angle between C1–C6/N1 (Fig. 1) and the fragment C9–C15/O2 is 28.5 (1)°. Atom C16 is displaced by 0.257 (4) Å from the mean plane through the latter fragment. The bond lengths and angles are in normal ranges (Allen *et al.*, 1987) and agree well with those in (5R,5'R)-2,2'-diacetamido-4,4'-di-*N*-acetyl-5'-(1-deoxy-2,3,4-tri-*O*-acetyl-D-erythritol-1-yl)-bis(1,3,4-thiadiazoline) mono-hydrate (Alho *et al.*, 2002).

Three intramolecular hydrogen bonds are present (Table 2) and, as a result, three pseudo-six-membered rings are formed (Fig. 1). In the crystal structure, the molecules are linked by intermolecular $N-H\cdots O$ interactions (Table 2), forming one-dimensional chains along the *b* axis (Fig. 2).

Experimental

A solution of *p*-phenylthiosemicarbazide (3.34 g, 20 mmol) in acetone (50 ml) was added dropwise to an acetone solution (50 ml) containing an equimolar amount of *p*-anisoyl chloride (2.52 g, 20 mmol) and ammonium thiocyanate. The solution was refluxed for

1 h and then poured into a beaker containing some ice blocks. The resulting white precipitate was filtered and washed with distilled water and cold ethanol, then dried under a vacuum. Crystals suitable for X-ray investigation were obtained by recrystallization from dimethyl sulfoxide (yield 85%, m.p 447.8–449.3 K).

Mo $K\alpha$ radiation

reflections

 $\theta = 1.7 - 25.2^{\circ}$ $\mu = 0.21 \text{ mm}^{-1}$

T = 298 (2) K

 $R_{\rm int}=0.049$

 $\theta_{\rm max} = 25.2^{\circ}$

 $h = -11 \rightarrow 11$

 $k = -18 \rightarrow 18$

 $l = -18 \rightarrow 27$

Block, colourless

 $0.48 \times 0.30 \times 0.20 \text{ mm}$

3073 independent reflections

2446 reflections with $I > 2\sigma(I)$

Cell parameters from 2397

Crystal data

 $C_{18}H_{19}N_3O_2S$ $M_r = 341.42$ Orthorhombic, *Pbca* a = 9.4042 (14) Å b = 15.603 (2) Å c = 23.143 (4) Å $V = 3395.8 (9) \text{ Å}^3$ Z = 8 $D_x = 1.336 \text{ Mg m}^{-3}$

Data collection

Bruker SMART APEX CCD areadetector diffractometer ω scans Absorption correction: multi-scan (*SADABS*; Bruker, 2000) $T_{\min} = 0.907, T_{\max} = 0.960$ 16783 measured reflections

Refinement

Refinement on F^2	$w = 1/[\sigma^2(F_o^2) + (0.0359P)^2$
$R[F^2 > 2\sigma(F^2)] = 0.063$	+ 2.015P]
$wR(F^2) = 0.124$	where $P = (F_0^2 + 2F_c^2)/3$
S = 1.25	$(\Delta/\sigma)_{\rm max} < 0.001$
3073 reflections	$\Delta \rho_{\rm max} = 0.19 \ {\rm e} \ {\rm \AA}^{-3}$
224 parameters	$\Delta \rho_{\rm min} = -0.25 \text{ e } \text{\AA}^{-3}$
H atoms treated by a mixture of	
independent and constrained	
refinement	

Table 1

Selected bond lengths (Å).

S1-C7	1.745 (3)	N1-C6	1.407 (4)
S1-C8	1.844 (3)	N2-C7	1.286 (3)
O1-C9	1.235 (3)	N2-N3	1.405 (3)
O2-C13	1.371 (3)	N3-C9	1.355 (3)
O2-C16	1.412 (4)	N3-C8	1.481 (3)
N1-C7	1.363 (3)		

Table 2	
Hydrogen-bond geometry (Å,	°).

$D - H \cdots A$	D-H	$H \cdot \cdot \cdot A$	$D \cdots A$	$D - \mathbf{H} \cdot \cdot \cdot A$
C5-H5···N2	0.93	2.42	2.894 (4)	112
C11-H11···N2	0.93	2.51	2.828 (3)	100
$C17 - H17A \cdot \cdot \cdot O1$	0.96	2.40	2.832 (4)	107
$N1 - H1B \cdots O1^{i}$	0.845 (17)	2.116 (14)	2.923 (3)	160 (2)

Symmetry code: (i) -x + 2, $y - \frac{1}{2}$, $-z + \frac{1}{2}$.

The H atom of the aniline fragment was located in a difference map and refined isotropically, with a bond length restraint of N-H = 0.844 (10) Å. The remaining H atoms were located in a difference map, repositioned geometrically and refined as riding, with C-H = 0.93–0.96 Å and with $U_{\rm iso}(H) = xU_{\rm eq}(\text{carrier atom})$, where x = 1.5 for methyl H and 1.2 for all other H atoms.

Data collection: *SMART* (Bruker, 2000); cell refinement: *SAINT* (Bruker, 2000); data reduction: *SAINT*; program(s) used to solve



Figure 1

The molecular structure of the title compound, (I). Displacement ellipsoids are drawn at the 50% probability level. Dashed lines indicate the intramolecular hydrogen bonds.



Figure 2

Packing diagram of (I), viewed down the *a* axis. The dashed lines denote the $N-H\cdots O$ intermolecular hydrogen bonds.

structure: *SHELXTL* (Sheldrick, 1997); program(s) used to refine structure: *SHELXTL*; molecular graphics: *SHELXTL*; software used to prepare material for publication: *SHELXTL*, *PARST* (Nardelli, 1995) and *PLATON* (Spek, 2003).

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